Year 7 Acids and alkalis fact sheet

Acids and alkalis	
	Lemon juice
1. Give an example of an acid	Vinegar
	Hydrochloric acid
2. Give an example of an alkali	• Soap
	Washing up liquid
3. Give an example of a neutral liquid	Water
Testing pH	
1. What scale do we use to test if a liquid is acid or alkali?	pH scale
	Put a drop of the liquid onto pH
2. Describe how to test the pH of a liquid	paper
2 M/hot colour do coido co with weivered	Compare the colour to the chart
3. What colour do acids go with universal indicator?	Red/ orange/ yellow
4. What colour do alkalis go with universal indicator?	Blue/ purple
5. What colour do neutral liquids go with universal indicator?	Green
6. What colour does a strong acid go with universal indicator?	Red
7. What colour does a strong alkali go with universal indicator?	Purple
8. What pH does an acid have?	Between 1-6
9. What pH does an alkali have?	Between 8 and 14
10.What pH does a neutral solution have?	7
11.What pH does a weak acid have?	5/6
12.What pH does a weak alkali have?	8/9
Indicators	
1. Name the chemical we use to test a liquid to see if it is acid or alkali	Indicator
2. Explain how to tell if a substance is a	It will turn a different colour in acid,
good indicator	alkali and neutral liquids

Ne	eutralisation	
1.	Describe how to neutralise an acid	Add an alkali You should test it with universal indicator to check when it gets to neutral
2.	Describe how to neutralise an alkali	Add an acid You should test it with universal indicator to check when it gets to neutral
3.	What do we call a reaction which starts with an acid and alkali and ends with neutral substances?	Neutralisation
4.	Write the general equation for neutralisation	Acid + alkali → salt + water
5.	List some useful neutralisation reactions	 Adding acid/ alkali to soil to make it neutral Treating bee/ wasp stings Indigestion tablets which neutralise stomach acid Making salts
M	aking salts	
1.	Are salts acid, alkali or neutral?	Neutral
2.	What type of reaction makes a salt?	Neutralisation
3.	A salt has 2 names. The first name is always a and the second name comes from the used to make the salt.	metal (first name) acid (second name)
6.	What type of salts are made from hydrochloric acid?	Chlorides
7.	What type of salts are made from nitric acid?	Nitrates
8.	What type of salts are made from sulphuric acid?	Sulphates
9.	Write a method to make a salt	 Mix the acid and base together Filter to remove unreacted solid Evaporate the water